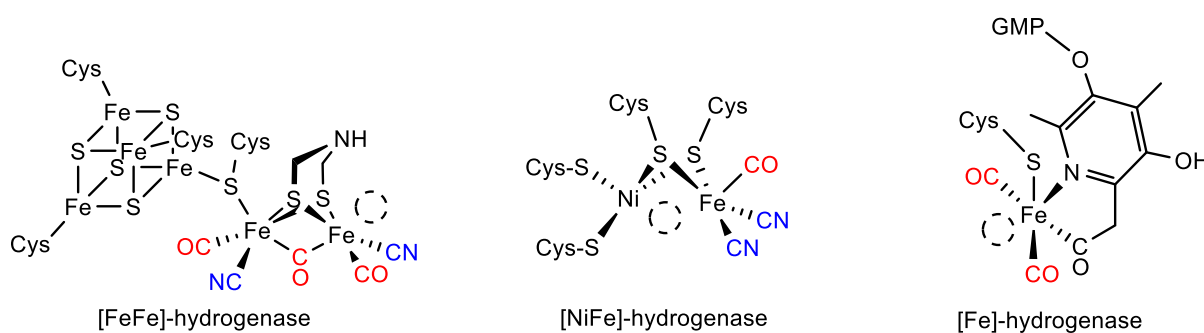


## The Biological H<sub>2</sub> Conversion: Pitfalls and Potentials of Hydrogenase Enzymes

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There is a general agreement that the building of a sustainable hydrogen economy, based on the combined development of photoelectrolyzers (for energy storage through water splitting into dihydrogen and dioxygen) and of fuel cells (for converting H<sub>2</sub> and O<sub>2</sub> back into electricity) in part relies on the availability of cheap, abundant, and efficient catalysts. The discovery of hydrogen-metabolizing enzymes, hydrogenases, and subsequent understanding of their structure and chemistry have been major breakthroughs towards novel catalytic strategies aimed at replacing platinum, currently used in the technological devices mentioned above.<sup>1,2</sup> Not only hydrogenases work with remarkably high catalytic rates at the thermodynamic equilibrium but also these metalloenzymes employ first row transition metals, nickel and/or iron, within their active sites (**Figure**).<sup>3</sup> The talk will give a general overview of the biological H<sub>2</sub> conversion, highlighting the role played by hydrogenases as well as pitfalls of ongoing research on these enzymes. Finally, a few examples showing promising achievements using hydrogenases will be illustrated.



**Figure.** Schematic representation of the H<sub>2</sub>-converting active sites found in hydrogenases. The dashed circle highlights the open coordination site used to bind H<sub>2</sub>. Cyanide (CN) and/or carbon monoxide (CO) ligands coordinate the Fe ions in all three hydrogenase classes.

### References:

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